

## Letter to the Editor

# Solution Chemistry Studies of Adriamycin—Iron Complexes Present *in Vivo*

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GOSÁLVEZ recently reported a new iron derivative of the antitumor agent adriamycin [1]. The lyophilized, freeze dried, mixture of three ferric chlorides and an adriamycin was called quelamycin. It has undergone clinical trials and aroused considerable research interest [2, 3].

We have used our established bio-inorganic techniques to investigate the binding constants for adriamycin-iron and other metals and considered which binding sites are involved. We also considered whether quelamycin dissociates into its components in plasma and whether adriamycin forms metal complexes *in vivo* [4-6].

The stability constants were determined potentiometrically and spectrophotometrically and are listed in Table 1.

Table 1. Log formation constants for adriamycin,  $-H^+$ ,  $-Fe^{3+}$  and  $-Cu^{2+}$  systems at 37°C

Species	Log $\beta$
AdH <sup>-</sup>	11.21
AdH <sub>2</sub> <sup>o</sup>	19.29
AdH <sub>3</sub> <sup>+</sup>	21.43
FeAd <sup>+</sup>	17.98
FeAd <sub>2</sub> <sup>-</sup>	29.03
FeAd <sub>3</sub> <sup>3-</sup>	33.41
FeAd(OH) <sup>o</sup>	14.69
CuAd <sup>o</sup>	12.72
CuAd <sub>2</sub> <sup>2-</sup>	19.27
CuAd <sub>3</sub> <sup>4-</sup>	23.21
CuAd(OH) <sup>-</sup>	5.74

I = 150 mmole dm<sup>-3</sup> [7].

Our results indicate that adriamycin has three good donor sites—the two-O<sup>-</sup> groups on the naphthacene rings in positions 6 and 11, and the sugar amino moiety. Of the metal ions studies, adriamycin had its strongest binding affinity for ferric iron ( $K \doteq 10^{18}$  mole<sup>-1</sup> dm<sup>3</sup>). Our studies involved adriamycin and ferric ions at equilibrium. Under these circumstances the 3:1 Fe(III)-adriamycin complex called quelamycin is not formed upon administering adriamycin to blood plasma whereas a calcium-adriamycin complex is probably formed.

Turning to non-equilibrium conditions, our data leads us to believe that quelamycin persists in plasma for several hr and should not be simply regarded as just a source of readily labile ferric ions and adriamycin. It may be regarded as a new agent rather than a simple derivative of adriamycin because the former is orally absorbed, has lower cardiotoxicity, and can pass the blood brain barrier [1-3].

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